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# The production of spelter as practiced at the Glendale Zinc Works

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18

84



THESIS FOR THE DEGREE

*of*

MINING ENGINEER

7591

*A. H. Standtler.*

18

84



### Subject.

The subject of this thesis is, "The production of Spelter as practiced at the Glendale Zinc Works."

### Description of the Zinc Works.

The "Glendale Zinc Works" are situated in the southern part of the city of St. Louis, on the line of the St. Louis Iron Mountain and Southern R. R., and among the iron manufacturing districts. They are the only zinc works now in operation, and produce a large quantity of spelter.

### Outline of process.

I will now proceed to give an outline of the process and treat of them separately going more into detail. The ores treated at these works are the "Blende" or "Zack" and the "Calamine". The "Blende" is crushed, and then roasted, while the "Calamine" is first calcined and subsequently roasted. The "Blende" and "Calamine" are mixed with bituminous coal, placed in a retort, to which an

intense heat is applied, the zinc distils and condenses in the condenser from which it is drawn.

### Details of process.

I will now proceed to give the details of the process, as carried on at these works, commencing with the mining of the ore, and ending in the production of spelter. For convenience I will treat of them under the following heads —

Ores.

Mining of ores.

Prep. of "Blende" before roasting.

" " "Calamine" after calcining.

Roasting.

Calcining.

Retorts.

Condensers.

Smelting.

Ores

"Blende"

This is a combination of Zn with S

having the formula in L, containing when pure 67.08% of Zn and 32.92% of L. A specimen from these works gave

Zn L =

Fe =

This ore of zinc occurs either in veins traversing the older rocks, or in floors and branches in more recent formations. The first mode of occurrence is perhaps the most frequent; but the more recent deposits are generally the most productive. This mineral occurs either massive, or in dodecahedra and octahedra forms. 2<sup>nd</sup> color, resin-yellow, to black (from imp.) Gravity about 4 1/2. The "Blende" is mined at Joplin, in the south-western part of the state of Missouri.

"Calamine".

The "calamine" is associated with "Blende", and is also obtained from Joplin. This ore is a <sup>hydrated</sup> silicate of Zn, having the formula  $ZnO \cdot SiO_2 + H_2O$ .

containing when pure 65.663% of ZnO,  
25.96% of  $\text{Li}_2\text{O}$ , and 8.38% of  $\text{H}_2\text{O}$ . A  
specimen from these works gave

ZnO = 63.212

$\text{Li}_2\text{O}$  = 22.289

$\text{H}_2\text{O}$  = 7.220

C  $\text{O}_2$  = 4.329

$\text{Al}_2\text{O}_3$  = 1.403

MgO = .094

CaO = .487 and  $\text{Fe}_2\text{O}_3$  = .917

It occurs in mammillated, massive,  
granular and crystalline forms, its  
color is yellow.

Mining of these ores at Joplin.

The ore as it comes from the mines  
of Joplin, is laid out on a platform,  
and the associated "Galena" is  
separated from it, with the aid of  
a hammer, it is then jigged, by a  
jigging apparatus of the following  
description - The jig consists of a  
box immersed in water, and by  
a proper arrangement is made to

move up and down rapidly, causing the lighter material, as earthy matter, rocks and etc., to rise, which are washed away, leaving the ore in the box. The washing is conducted in a series of gigs. The ore is now ready for shipment, the cost per ton at the present time is \$21.50 for the "Blende", and for Calamine \$17.

#### Prep. of "Blende" before roasting.

The gigged "Blende" is crushed between steel-rollers, it should pass through a 36 in. mesh, this being its proper fineness. The steel-rollers can crush from 30, to 40 thousand pound per day. They employ 2 men, one tends to the crushing, while the other is employed to run the engine; it is thence taken to the roasting furnaces. They receive resp. 1.50 and 2.75.

#### Prep. of Calamine after calcining.

In the case of the Calamine, the

ore is first calcined, and subsequently crushed in exactly the same way as in the case of the Blende. They crush about the same amount of ore.

### Roasting.

Before Zn. can be extracted from the "Blende", it must be roasted, that is it must be converted to the oxide, this is done by roasting the Zn. S., driving off the S in the form of  $S O_2$ , leaving it as Zn O. The roasting at these

works is performed in double-reverberatory furnaces, having an upper, and lower department, 16 working down, 8 above and 8 below, a high chimney, about 50 ft. <sup>in</sup> height, a hopper and a fire-place, for description of furnace and dimensions see Figs I and II.

### Working of a charge.

I shall take a single furnace, and go into the detail of roasting.

They generally put a quantity of the crushed Zn S. on top of the roaster,



allowing it to remain there some time. These roasts are transferred to the upper department, through a hopper in front of door number I, it <sup>remains</sup> there some time, being frequently <sup>stopped</sup> by means of iron paddles, it is now removed to door no II, while a fresh supply is put in number I, number II is moved along to number III, and number III to number IV, at number IV there is an opening, allowing the partially roasted "blende" to be transferred to the lower department. Then moved along to door number V, where it is tested to see if any sulphur remains, if the workmen sees that there is no more than 2% remaining, he draws it from the furnace, and <sup>is</sup> removed to the smelting furnaces. Each single furnace roasts from 6,000 to 6,200 yds. day of 12 hours, 12,000 yds. 24, employing experienced workmen, receiving .06 yds 100 lbs, They have 3 double-roasters.

roasting on an average 70,000 lbs.  
per day of 24 hours.

### Calcining.

Calcining is resorted to, in the case of Calamine. The calcining is done in an ordinary lime-kiln. The object being to drive off the water contained in it, 80,000 lbs are thrown in, the size being, as large as a doubled fist, and calcined for from 4 to 5 days, it is then crushed, and put into sheds.

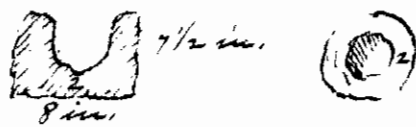
### Retorts.

As retorts are used in smelting the ore, I will first give the manner in which they are made. The company has a department for making retorts and condensers aside of the crushing department. They employ 5 hands, very skilled in their trade, in fact all the men employed, are men of experience, having had much practice.

## Making Retorts.

The retorts are made of crushed fine clay and old broken retorts. The fine clay and old broken retorts are mixed together, moistened with water and tempered, when this is done, the one who attends to this, rolls it in long round rolls, of about  $3\frac{1}{2}$  ft. in length and 1 inch in thickness.

The bottom of the retort is first made, by taking out a bunch of the material, rolling it into the form of a small cylinder and giving it a depression, it takes the form as shown in the figure below.



when this is completed, the rolls are rolled around, the bottom up to  $\frac{1}{2}$  its intended height, being 29 inches. The outside is scraped smooth, with a piece of wood, a piece of cloth is now wrapped around the  $\frac{1}{2}$  retort. is put

in its mould, which is a cast iron  
shell, having a cylindrical form,  
opened and closed by hinges, fig  
III. represents the mould. The inside  
is next scraped smooth with a  
scraper, a rammer is then plunged  
into the hollow of the cylinder,  
which is an instrument shown in  
fig. IV. a borer is now placed  
into the hollow of the cylinder, and  
turned around, this making the  
sides compact and giving it  
proper diameter, when this opera-  
tion is completed it is beaten, with  
the beater, it being a stick shod  
with iron, fig V. represents it, the  
last operation is smoothing  
the inside, or as they term it,  
polishing. The other  $\frac{1}{2}$  is built upon  
this, in exactly the same way,  
of course there is no bottom made.  
The retort is 49" long, 10" in external  
diameter and 8" in internal. They are  
then taken to the drying house.

which is heated from the waste-steam of the engine employed in the steel-rollers, kept there from 4 to 6 weeks, from thence they are removed to the kiln, burnt there from 12 to 15 hours, and are then ready for use. They employ 4 men, receiving .17 \$ to a piece, they make from 17 to 20 per day of 12 hours.

### Condensers.

The condensers are made of the same material as the retorts.

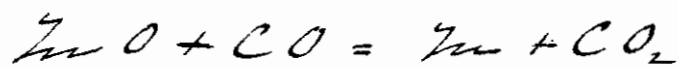
### Making Condensers.

In making the condensers the clay is annealed, a large bunch is placed in the bottom of a mould, it being made of cast-iron, having the shape of a hollow truncated cone, a rammer is now rammed into the middle of it, this gives it its proper form, the mould with the condenser is then carried to

its proper place, he taps it, which loosens it from its mould, they are next dried for a week or two, burnt, and are ready for use. They employ two men, receiving .01¢ a piece, they make from 250 to 300 per day of 12 hours. The condensers are 1 1/2 ft long.

### Smelting.

The final treatment is the smelting, from it we obtain the metallic zinc. This is effected by the reaction of carbon on the zinc oxide, by the aid of heat; the reaction is shown below.



### Furnace.

The furnaces employed at these works, are double-furnaces, 16 by 16 by 14 ft. The back of the opening, meaning by the back the wall between the front walls of the double furnaces, which hold the closed end of the retorts, is slightly inclined in the direction at

as shown in fig. VI. The fire place B, is placed beneath the surface of the ground, the flame and heated air enter the interior of the furnace through  $\gamma$  opening c. in each single furnace is placed 108 retorts, 11 in a row, and 10 rows high, on the open face of the furnace, are arranged 10 plates of cast-iron, protected on the inner side by fire-clay; these are destined for the support of the condensers.

The height of the steps at the back of the furnace and that of the iron plates in front of the openings, is so arranged as to give the retorts a slight inclination.

#### Working of a charge.

Assume that a charge has been taken out, and the retorts and condensers have been cleaned. The ore which is in front of the furnace consists of  $\frac{2}{3}$  blende  $\frac{1}{3}$  calamine and pulverized bituminous coal,

the workmen throw in in the charge with semi-cylindrical shovels, a shovel hold about 10 lbs of the mixture. 50 lbs. are thrown in, the condensers are now coated to the retorts with fire clay, the fire urged; at first there is a dense black smoke, which last about 15 minutes. Then a blue flame appears at the openings of the condensers. At the expiration of a further period the brilliancy of the combustion is considerably increased, and the flame at the same time assumes a greenish-yellow tint, and gives off copious white fumes, at this period, they place at the ends of the condensers, wrought iron adapters to catch as much ZnO as possible, at the expiration of 2 hours, the workmen remove the adapters, tap them, and save the ZnO for future use. When



This has been done, an assistant holds a large iron ladle under the beak of each condenser, at the same time that the foreman draws out into it with an iron scraper the distilled zinc <sup>which is put into molds, mold holding 5 lb.</sup> The manipulations are repeated at intervals of two hours, in about 12 hours time the distillation has commonly terminated. They obtain 20 lbs of Zn from 50 lbs of ore, employing 3 men at each single furnace; they work day and night, they have 4 double furnaces, although they generally only run three. At the expiration of 2 months they deem it necessary to repair a furnace, so that while three are running the other is being repaired, they men receive from 2.50 to 3.50, depending upon the seniority of position. The retorts last from 1 to 2 months, the top men, last longer.

The following is analysis of the spectra  
produced at these works.

$$\text{Fe} = 99.263054$$

$$\text{Pt} = .542891$$

$$\text{Cd} = .164365$$

$$\text{Zn} = .009353$$

$$\text{Sn} = .007097$$

$$\text{Cu} = .006152$$

$$\text{SiO}_2 = .002320$$

$$\text{As.} = .001768$$

$$\text{Bi} =$$

$$\text{S} = \left\{ \text{Trace.} \right.$$

$$\text{C} =$$

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$$100.00000$$

The present cost of spectra is  
\$4.37½ per hundred pounds.

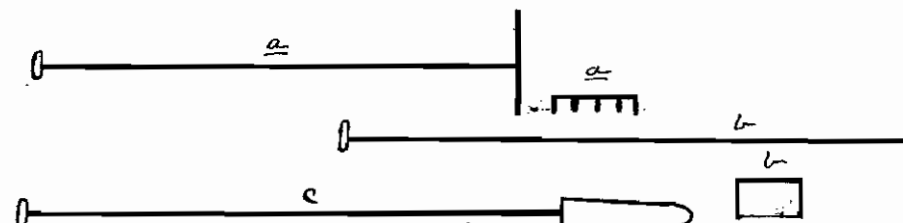
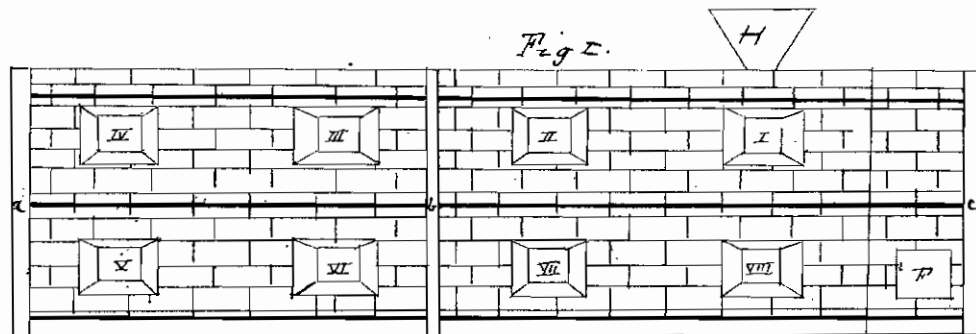


Fig I. represents a longitudinal section of roaster.  
 I, II, III, etc. are working doors.  
 a, b, c are bands of iron.  
 P is the fire-plate.  
 A single furnace is 30 by 9 by 8.  
 H is the hopper.

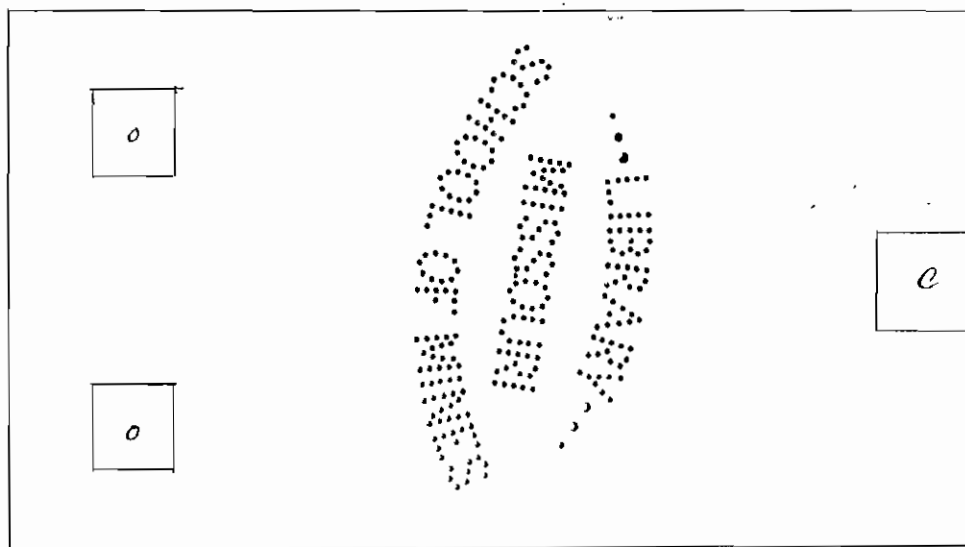
Figs a, b and c are tools used.  
 Fig a is a rake.  
 Fig c is a rafter.  
 Fig b is used to draw the roasted ore from furnace.

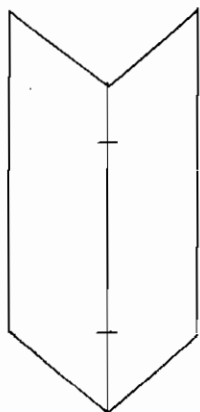
Fig II rep. horizontal section through top floor.

o, o are door through which the ore is transferred to lower department.

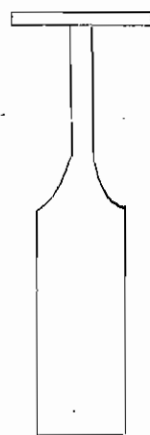
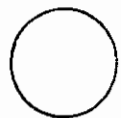
c is the flue.

Fig II.

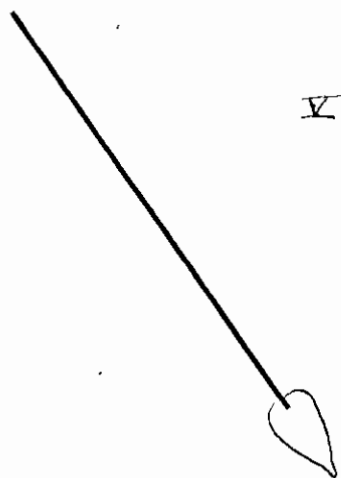




*Fig III.*



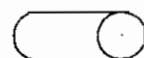
*VI.*  
*a borra.*



*V.*



*Fig IV.*



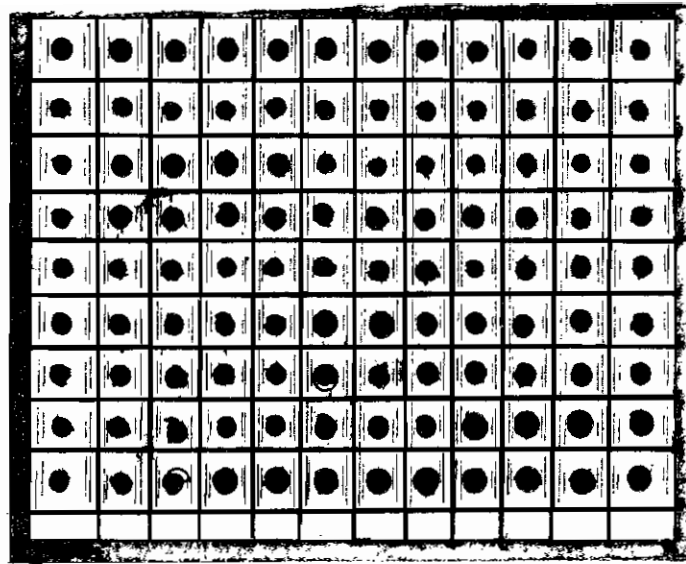


Fig VI. Represents horizontal and vertical sections.  
 c, c c are the condensers.  
 w. b. b. " " returns.

